# INVESTIGATION OF TITANIUM COMBUSTION CHARACTERISTICS AND SUPPRESSION TECHNIQUES

FIRE PROTECTION BRANCH
FUELS AND LUBRICATION DIVISION

FEBRUARY 1976

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This test program studies the but flow conditions. The flat plate ti from an electrically heated ignitor combustion of a single sample are tests with steady state burning.  Argon gas is shown to be a feasi	arning charactering tranium samples at a flow conductor and the samples at the sa	stics of titanium under air are ignited by molten titanium ditions that support sustained burn rate is measured on all
fire. Quick injection of a suffici	ient amount of ar	gon gas to maintain a 60%

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### 20. ABSTRACT - Continued

concentration by volume of argon results in quick suppression by oxygen depletion. Carbon dioxide  $(CO_2)$ , a common fire extinguishing agent, is shown to sustain titanium burning at an accelerated rate.

The ultraviolet (UV) radiation emitted by burning titanium is shown to be of a sufficient intensity for existing UV fire detectors to detect at reasonable distances.

### FOREWORD

This report was prepared by Duane G. Fox of the Fire Protection Branch, Fuels and Lubrication Division, Air Force Aero-Propulsion Laboratory (AFAPL/SFH). The work reported herein was performed under Project 3048, "Fuels, Lubrication, and Fire Protection," Task 304807, "Aerospace Vehicle Fire Protection," Work Unit 30480773, "Aircraft Fire and Explosion Prevention." This work was performed at the request of the Components Branch of the Turbine Engine Division in support of Work Unit 30661005, "Compressor Rotor Rub Test." Test conditions and requirements were supplied by Mr. Charles W. Elrod, AFAPL/TBC.

This report covers research accomplished in-house from January 1974 to March 1975.

The author appreciates the assistance received from Mr. Jon R. Manheim, AFAPL/SFH, in designing the combustion chamber and extinguishing test hardware and in developing the experimental tests. Special thanks are given to the following individuals: Messrs. Peter Danelak, Harvey Reeves, Glen Boggs, and Robert Esch of AFAPL for their invaluable help in the execution of the experiments.

This report was submitted by the author July 1975.

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### SECTION I

# INTRODUCTION AND SUMMARY

# A. INTRODUCTION

This test program was initiated to study the burning characteristics of titanium under specified flow conditions and to find a technique for extinguishing an on-going titanium fire in a test facility. The work was accomplished prior to the operation of a full-scale, single stage compressor test facility.

This program had two primary objectives: (1) Tests were conducted to determine what conditions (air temperature, air pressure, and air flow) are required for sustained combustion on a single compressor blade and representative flat plate sample. The burning rate was determined for all cases of sustained combustion. (2) Suppression studies were conducted to determine what concentration of an inert gas such as argon is required to extinguish a titanium fire.

### B. SUMMARY

It was found that the burning characteristics of titanium samples are not strongly dependent on air flow temperature or pressure within the limits established in this program (121°C < T < 399°C, 448 kPa < P < 1138 kPa). The initiation of sustained burning and burning rates are more dependent on the sample shape, thickness, and relative position to the air flow. Limited testing of a  $\beta$  alloy of titanium indicates that material composition does affect the burning characteristics.

Measurement of the ultraviolet (UV) radiation emitted from the burning titanium indicates that the UV emitted from a  $2.54 \times 7.62$  cm sample is at least an order of magnitude greater in intensity than from a 5-inch diameter hydrocarbon fuel fire. Utilization of a UV fire detector for detecting a titanium fire is thus feasible.

It was shown that, for the test conditions studied, an argon gas concentration of at least 60% is required to extinguish a burning titanium sample. The argon dilutes the oxygen concentration to a level that will not support sustained combustion. Since substitution of argon for air can be done rapidly without significantly changing the total air flow through a test device, this extinguishment technique is applicable to turbine engine compressor test facilities where titanium combustion presents a hazard. The argon concentration must, however, be maintained until either the molten material cools sufficiently to prevent re-ignition or until the air flow is reduced so that sustained burning can not continue.

While steady-state burning data was obtained for the single sample, direct extrapolation to a rotating environment with complex air flow patterns such as exists in a turbine engine compressor is not possible. At best the steady-state burning data obtained can be used in computer modeling of the complex solid combustion phenomenon. The critical factor in achieving sustained burning following ignition and localized burning is the air flow and how it removes oxidized material from the surface. This phenomenon was not thoroughly studied in this effort.

# C. ADDITIONAL INVESTIGATIONS REQUIRED

Although the testing performed provides baseline data on the burning characteristics of titanium in air flow, additional testing should be

conducted to more fully characterize and define the effects of the ignition source and the air flow over the sample. The effects of a stacked sample array simulating an actual compressor also need to be adequately defined. This information will be required when the combustion phenomenon is modeled. Tests should also be conducted with other types of extinguishing agents.

### SECTION II

### EXPERIMENTAL EQUIPMENT

### A. TEST FACILITY

The tests were conducted in a test facility located at the Air Force
Aero-Propulsion Laboratory at Wright-Patterson AFB. The facility was
developed to test turbine engine combustors. The facility provides air
at a regulated pressure, temperature, and flow. A simplified schematic
indicating the components of importance to the titanium combustion tests is
shown in Figure 1. The control instrumentation is shown in Figure 2 and
Figure 3. The overall test facility is shown in Figure 4.

The air is supplied by piston compressors and is then heated to the required temperature by a furnace. Pressure is maintained at a prescribed value in the test chamber by a feed back controller which opens or closes a bleed off valve. The flow is regulated by opening or closing a plug type orifice at the end of the air flow section. The plug orifice was operated by a remote switch in the control room.

The test sequence used was to first set the desired air temperature and pressure and then regulate the plug to get the required air flow. Closing the plug, for example, decreases the flow through the test chamber and increases the amount of bleed off.

The flow is determined by measuring the differential pressure ( $\Delta P$ ) across a two-inch diameter venturi and using the standard equation which relates the  $\Delta P$  to the mass flow in pounds of flow per second. For ease of operation, tables were tabulated by a computer. By entering the temperature, static

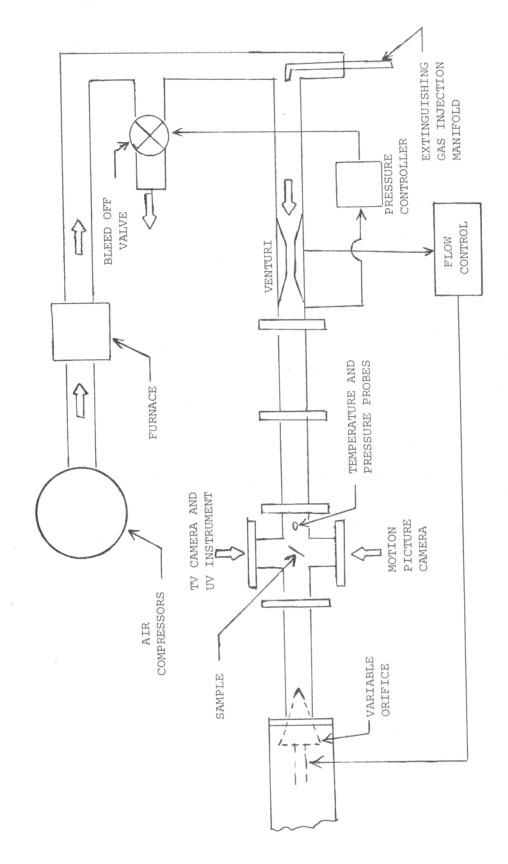


Figure 1. Titanium Test Facility Schematic

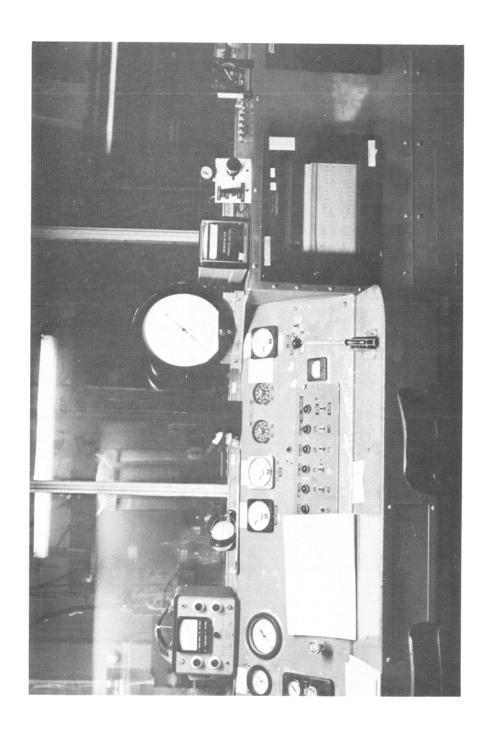


Figure 2. Air Supply Control Instrumentation

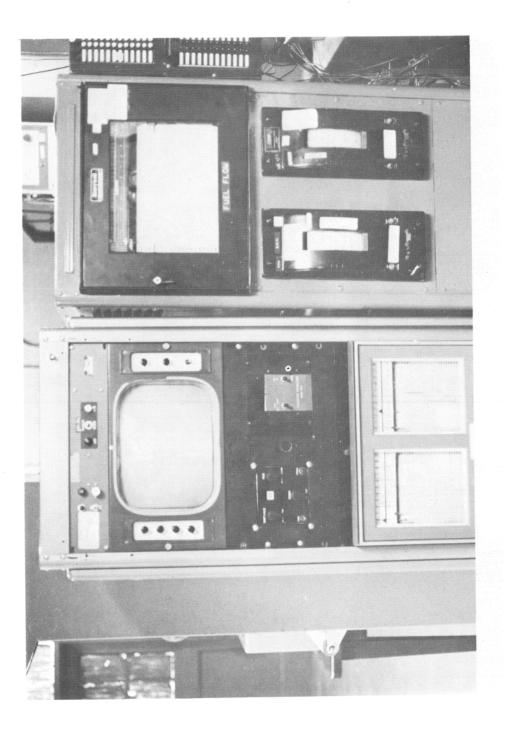


Figure 3. Air Supply Parameter Instrumentation

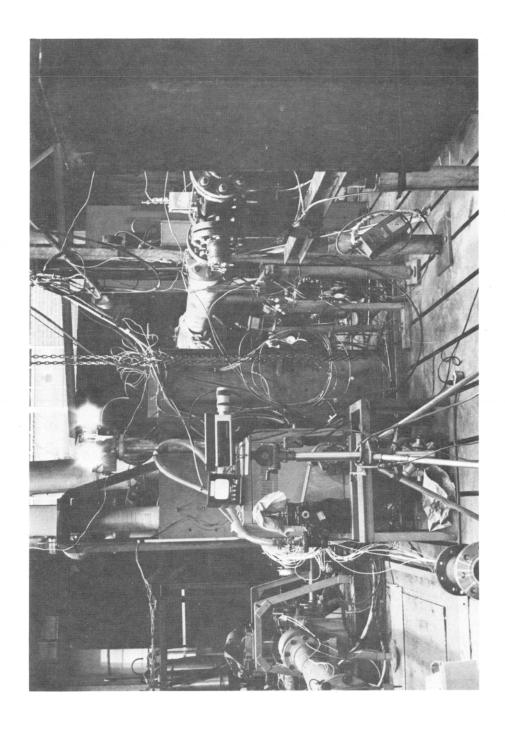


Figure 4. Photograph of Titanium Test Hardware

pressure, and differential pressure, the table gives the mass flow. The flow is reported in both Kg/sec and lb mass/sec in this report. Air velocity in the chamber is calculated from the pressure, flow, and temperature and is reported in both meters/sec and feet/sec.

### B. COMBUSTION TEST CHAMBER

The test chamber was designed to use readily available materials in order to shorten fabrication time. The original chamber, prior to a few later modifications, is shown in Figure 5. The first 24-inch long section of pipe isolates the flow measuring venturi from the test chamber. The chamber is a standard 300 pound pipe cross. One leg of the cross contains a water cooled jacket which houses a 7.6 cm (3 inch) diameter, 1.27 cm (1/2 inch) thick quartz window. This window provides access for television camera coverage and also permits measuring the ultraviolet (UV) radiation emitted from the titanium flame. The UV radiation is of interest because it is a likely technique for detecting a titanium fire.

The other cross leg contains a water-cooled jacket which houses a 10.2 cm (4 inch) diameter, 4.4 cm (1.75 inch) thick tempered pyrex glass window. This window provides viewing access for high speed motion picture photography. The window jacket is cooled to prevent the glass from weakening at the higher temperatures. The cooling, however, causes a temperature gradient across the glass which results in the glass fracturing at a temperature near 399°C (750°F). The window was used satisfactorily for tests at 121°C (250°F) and 260°C (500°F). The glass window was replaced by a steel plate for the tests at 399°C (750°F). In these tests, the motion picture camera and TV camera both view through the smaller quartz window by the use of a partially reflecting mirror arrangement which is shown in Figure 6. This scheme proved adequate; however, alignment is more critical through the smaller window.

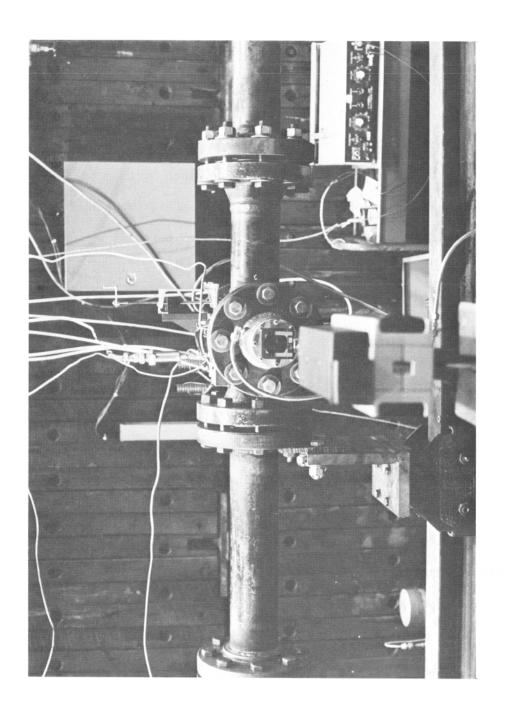


Figure 5. Test Chamber

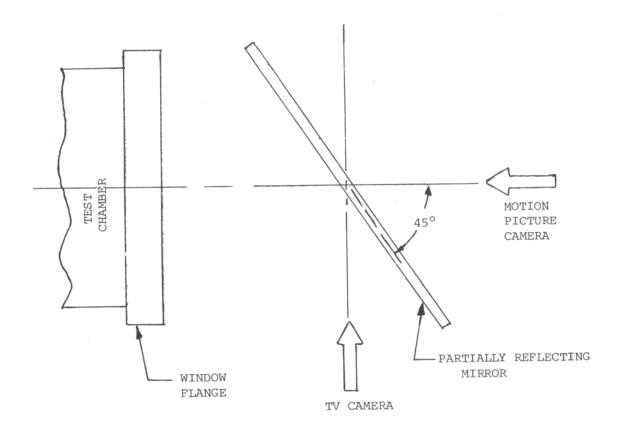
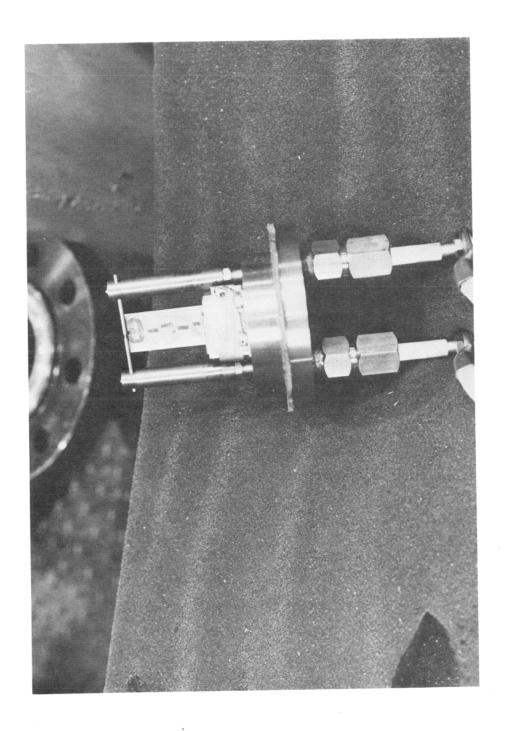


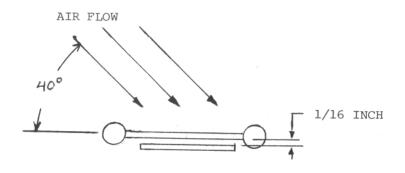
Figure 6. Mirror Arrangement for Visual Observation

The sample holder and igniter are mounted on a flange which bolts into the bottom center of the test section. This arrangement permits a fast change of test specimens. The sample holder and igniter are shown in Figure 7. The sample is held in place by boron nitride blocks which are fastened to the flange by stainless steel brackets. The boron nitride blocks keep the sample from burning past the holder. Since boron nitride is a high temperature material, it works satisfactorily at the high ambient temperature and is not significantly damaged by the molten titanium.

The igniter shown in Figure 7 is a 0.23 cm (0.090 inch) diameter titanium rod which is machined to 0.15 cm (0.060 inch) diameter at the center for 0.64 cm (0.25 inch) length. This forces the igniter to burn first at the center. Without the narrowed section, the igniter will usually burn first at one end or the other because of the strain produced at the connection point. This igniter proved to be unreliable at high air flow. Analysis of the high speed motion picture film revealed that the igniter was blowing over the top of the sample after becoming soft prior to melting.

The igniter was modified to a 6.3 mm x 1.6 mm x 7.62 cm long (0.25 inch x 0.062 inch x 3 inch) piece of titanium which is positioned evenly with the top edge of the sample, as shown in the illustration in Figure 8. This igniter is notched approximately 1.6 mm (1/16 inch) deep on both sides at the center. This igniter proved to be reliable and was used for the tests described in this report. Electric current from a 200 ampere, 16 volt, 60 Hertz transformer passes through the electrical fittings to the igniter holder. All conductors in the igniter circuit except the igniter are made from copper.





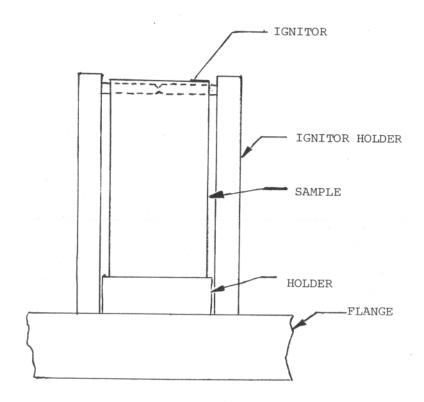


Figure 8. Sample and Ignitor Arrangement

### C. SUPPRESSION HARDWARE

The suppression tests primarily involved injecting argon gas into the air flow upstream of the burning titanium sample and noting effects on the burning sample. CO<sub>2</sub> gas was also used in a few tests. The injection manifold is illustrated in Figure 9. This manifold injects the gas as illustrated in Figure 1. The manifold is pressurized up to the solenoid valve by a high capacity regulator which is manifolded to twelve, Size A argon cylinders. The complete argon injection system schematic is shown in Figure 10. The same hardware was also used for the CO<sub>2</sub> studies except that only six bottles were employed.

The argon temperature and pressure are measured in the injection manifold. The injection system was calibrated by sampling the flow stream near the sample. This procedure will be detailed later.

### D. INSTRUMENTATION

The air system control instrumentation consists of a pressure gauge readout of the static wall temperature, a strip chart recorder output of the differential pressure ( $\Delta P$ ) across the venturi, and a thermocouple meter output of the air stream temperature. These instruments were used for adjusting the air flow conditions in the test chamber.

The test chamber parameters were recorded on a chart recorder (shown in Figure 11) so that changes and transients could be observed. The conditions during the burning tests were found to be stable and thus did not actually require time recording. The argon suppression tests, however, do involve rapid changes of temperatures and pressures and the recorded data is required for analyzing the test results.

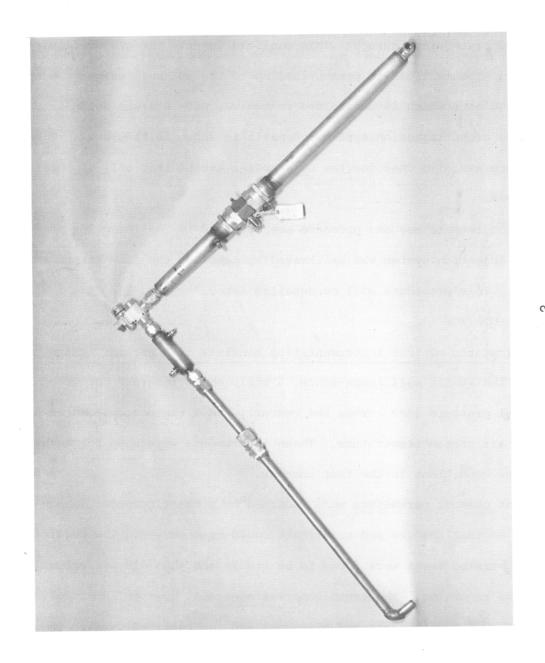


Figure 9, Argon and CO<sup>2</sup> Injection Manifold

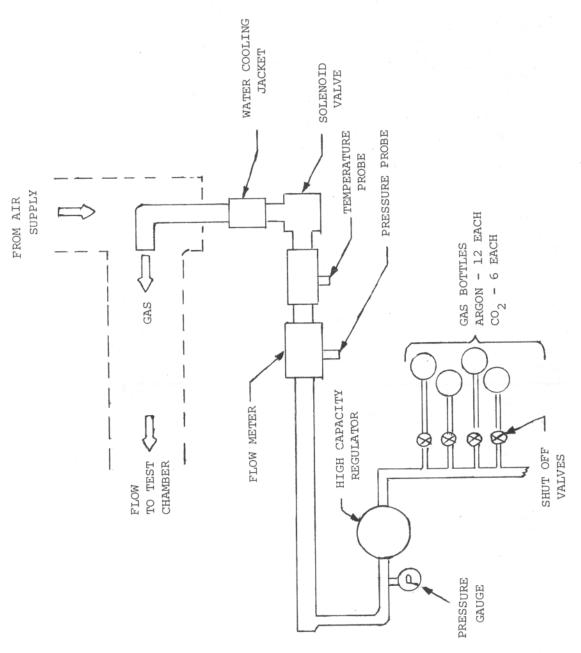


Figure 10. Argon and  ${\rm CO}_2$  Supply System Schematic

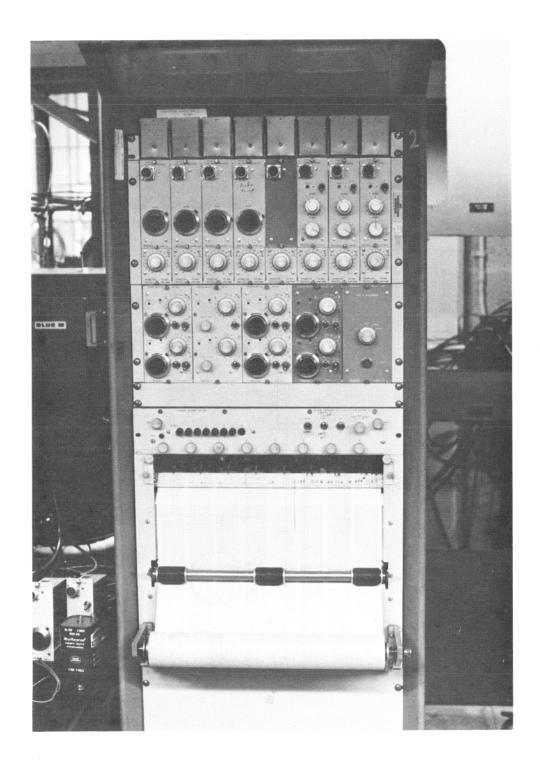


Figure 11, Data Chart Recorder

The air temperature is measured by an exposed junction Cr-Al thermo-couple which is located about 6.4 mm (0.25 inch) from the wall in the nozzle just upstream of the sample. The exposed junction provides sufficient response during the argon injection tests.

The chamber static pressure is measured at the wall just upstream of the sample. The pressure transducer is located near the chamber and provides a sufficient frequency response to record transients in the pressure.

Two event markers are used so that both the time of ignition and the time of argon injection can be correlated with the pressure and temperature traces.

The argon temperature is measured near the exit of the flow meter.

Measurement of the argon flow by using a turbine type flow meter proved to be unsuccessful because the readings were difficult to interpret. The actual argon concentration was determined by sampling the air stream near the sample and analyzing the mixture for oxygen, nitrogen, and argon. Samples were taken at the six required test conditions. The specific details of this procedure will be discussed in the section on extinguishing tests.

The ultraviolet emission tests were made with a spectroradiometer system that will be described in detail in Section VI.

The tests are properly sequenced by the use of a 12-step sequencing programmer which has a dwell time on each step that is adjustable from 0 to 10 seconds. This programmer (shown in Figure 12) is used to activate the motion picture camera, TV camera, sample igniter, argon injection valve, sample valves, and provide sync signals for the recorders and cameras. Once initiated, the test is automatic but does have manual override on some of the functions.

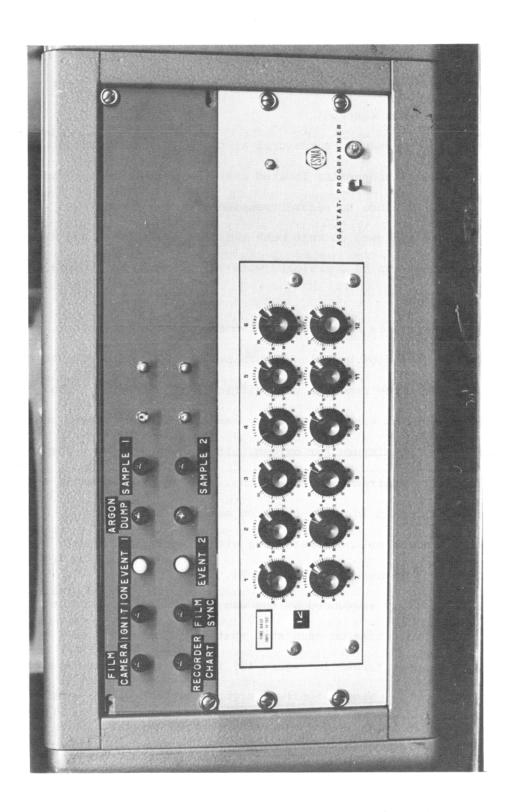


Figure 12, Test Sequence Programmer

### SECTION III

### COMBUSTION TESTS

### A. TEST DESCRIPTION

The test procedure used was to first install a sample in the test chamber and then bring the air flow, pressure, and temperature to the desired values. The test was then initiated by starting the test sequencer. The test was monitored on the TV system and a determination made on whether the ignition was normal and whether sustained combustion occurred. In addition, the sample was analyzed visually after removal from the test chamber. These first hand procedures were successful in determining if the test was satisfactory for most tests. A more detailed analysis was later conducted by looking at the high speed motion pictures. This analysis showed that a few tests which were initially thought to be good were, in fact, not valid because of ignition difficulties. These tests were then excluded from the study.

If the TV viewing and sample analysis showed that the ignition was not normal, the test was repeated. If sustained combustion occurred, the air flow for the next test was increased and if non-sustained combustion occurred, the air flow was decreased. Eventually, the critical value of air flow was found that separated the non-sustained combustion and the sustained combustion regions. This sequence was repeated at all combinations of the three pressures (448, 793, 1138 kPa) and the three temperatures (121°C, 260°C, and 399°C).

Tests were conducted with two sample thicknesses (0.06 cm and 0.16 cm).

In addition, compressor blades were tested at some of the pressures and temperatures. The limited number of blades and facility test time available did

not permit testing the blades over the complete range of temperature and pressure. Enough tests were conducted, however, to allow some comparison of results.

The combustion tests were designed to define the air flow conditions that would support sustained combustion on a sample ignited on the edge by molten titanium. This is both a function of the air flow conditions around the sample and the ignition source. An insufficient amount of energy in the ignition source will fail to ignite a sample even though the airflow conditions are amenable for sustained combustion. The effect of the ignition source was not thoroughly studied in this test program, however, the high speed motion pictures of the ignition and sample were studied to determine the characteristics of the ignition source.

If the molten material ignited the sample along the top edge and the air flow was correct for sustained combustion, the sample would burn completely to the sample holder. Some burn patterns produced an even, horizontal burn down the sample. Other burn patterns were more complex and resulted in the flame burning down the leading or trailing edge first and then burning forward or rearward into the sample. This was more prevelant with the actual compressor blades because the blade edges are thin and burn more readily than the center portion.

As the air velocity on other tests was increased, a point would be reached such that the sample would start to burn at the point that the molten titanium from the igniter impinged on the sample but would soon stop burning (usually within a few mm, but occasionally as much as one-half cm). Since the sample would burn a short distance and then stop, it was assumed that sufficient energy was present to ignite the sample.

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Several additional tests were conducted to establish the effectiveness of the ignition source. A steady burning was established on a sample by igniting it in air flow conditions that support sustained combustion. The air flow was then changed to a condition that had been established as a non-sustained burning condition. The sample stopped burning immediately after the air flow was changed. These tests further verify that the ignition source is sufficient to ignite the sample if the air flow is correct for sustained combustion.

The air flow required for sustained combustion is also a function of the angle of the sample relative to the direction of air flow. The angle used in these tests is 40 degrees, which is typical for a compressor blade. The effect of varying the angle between the sample and the air flow was not evaluated in this series of tests.

The following five samples were used in the combustion tests:

1. Sample A - Size: 2.54 cm x 7.62 cm x 0.06 cm

 $(1" \times 3" \times 0.025")$ 

Material: Titanium Alloy

6% Aluminum, 4% Vanadium

2. Sample B - Size: 2.54 cm x 7.62 cm x 0.16 cm

 $(1" \times 3" \times .064")$ 

Material: Titanium Alloy

6% Aluminum, 4% Vanadium

3. Sample C - Compressor Blade

6% Aluminum, 4% Vanadium (thickness less than Sample D)

4. Sample D - Compressor Blade

6% Aluminum, 4% Vanadium (thickness greater than Sample C)

5. Sample E - Size: 2.54 cm x 7.62 cm x 0.11 cm

Material: Titanium Alloy

 $\beta$  structure

### B. TEST RESULTS AND DISCUSSION

The test results are tabulated in Table 1 through Table 9. The critical value of the air flow for supporting sustained combustion is tabulated in Table 10. The critical value is determined by looking at all tests of one sample at a fixed set of air flow conditions and estimating the break point between the sustained and non-sustained burning regions. This is not necessarily the midpoint between the data points, but is the result of analyzing the individual tests and considering factors such as ignition. It should be apparent that this value could vary somewhat due to the interpretive factors involved in making the determination. This value is, however, the best that can be obtained from this series of tests. The results are sufficient to establish trends in the burning characteristics over the temperature and pressure range of interest in this study.

Sample A generally burned at a higher air velocity than Sample B, which is expected because of the difference in thickness. The effect of an increase in air temperature is an increase in the air velocity at which sustained combustion can occur, as illustrated in Figure 13. The effect of pressure varies, as illustrated in Figure 14.

The data on Sample C is presented in Figure 15. It should be noted that two data points are not actual break points between non-sustained and sustained burning, but only measured data points in the sustained burning region. There were not sufficient tests conducted to determine the actual break point. The blade shows a definite effect of temperature and

TABLE 1

TITANIUM COMBUSTION TEST RESULTS, SAMPLE A

Summary of Test	Sustained combustion, entire sample burned Ignition was marginal, only slight combustion of sample Sustained combustion, sample burned about 75% Sustained combustion, sample burned about 85% Slight combustion on top edge	Sustained combustion, entire sample burned Slight combustion on top edge Partial combustion on top edge Partial combustion on top edge Sustained combustion, entire sample burned Sustained combustion, entire sample burned Partial combustion on top edge and leading edge Marginal ignition, igniter positioned low on sample Partial combustion on top edge	Sustained combustion, entire sample burned
Burn Rate cm/min (inch/min)	6.26	7.23 7.26 Not measured	Not measured
Air Velocity m/sec (ft/sec)	107 203 162 162 176	154 231 193 179 156 133 162 146 147	136
Air m/sec	32 62 49 64 54	44 10 10 10 10 10 10 10 10 10 10 10 10 10	41
Air Flow Kg/sec (lbm/sec)	11.22	00000000000000000000000000000000000000	2.5
Air   Kg/sec	0.36 0.68 0.54 0.59	0.91 1.36 1.13 1.04 0.91 1.13 1.22 1.22 1.22	1.13
Pressure Pa (psia)	65 65 65 65	114 114 113 113 113 165 163 162 162	162
Pres	455 448 448 448 448	786 786 779 779 779 1138 1124 11124 1117	1117
Test #	8AK01 8AK02 8AK03 8AK04 8AK05	8AK06 8AK07 8AK08 8AK09 8AK10 8AK11 8AK13 8AK13 8AK13	8AK16

Sample: 2.54 cm x 7.62 cm x 0.06 cm Titanium Temperature: 121°C  $\,$ 

TABLE 2

TITANIUM COMBUSTION TEST RESULTS, SAMPLE A

Summary of Test	Partial combustion on top edge Partial combustion on top edge Partial combustion on top edge Sustained combustion, entire sample burned Sustained combustion, entire sample burned	Partial combustion on top edge Marginal ignition, not considered a valid test Sustained combustion, entire sample burned Sustained combustion, entire sample burned Sustained combustion, entire sample burned	Sustained combustion, entire sample burned Sustained combustion, entire sample burned Partial combustion on top edge Sustained combustion, entire sample burned Partial combustion on top edge	Sustained combustion, entire sample burned Partial combustion on top edge Partial combustion on top edge Slight combustion on top edge Slight combustion on top edge
Burn Rate cm/min (inch/min)	5.17 Not measured	6.6 Not measured 8.3	9.6	17 6.7
locity (ft/sec)	360 284 203 133 135	206 154 115 134	107 134 188 161	155.6 175 175 171 171
Air Velocity m/sec (ft/se	110 86 62 41 41	63 47 35 41	33 41 57 49 50	47.4 53.3 53.3 52.1
Flow (lbm/sec)	11.00	1.5 2.0 1.5 1.75	0.	2.0 2.25 2.25 1.25 1.25
Air Kg/sec				0.90 1.02 1.02 0.57
Pressure Pa (psia)	61 62 65 66 65	64 114 115 115	164 164 164 162	113 113 113 64 64
Pre				779 779 779 441
Test #	8AM01 8AM02 8AM03 8AM04 8AM05	8AM06 8AM07 8AM08 8AM10	8AM11 8AM12 8AM13 8AM14 8AM15	8AN06 8AN07 8AN08 8AO10

Sample: 2.54 cm x 7.62 cm x 0.06 cm Titanium Temperature:  $260^{\circ}\text{C}$ 

TABLE 3

TITANIUM COMBUSTION TEST RESULTS, SAMPLE A

Summary of Test		Sample failed, not a valid test	Sustained combustion, entire sample burned	Slight combustion on top edge	Partial combustion on top edge	Partial combustion on top edge	Sustained combustion, entire sample burned	Sustained combustion, entire sample burned	Partial combustion on top edge	Partial combustion on top edge	Sustained combustion, entire sample burned	Partial combustion on top edge	Slight combustion on top edge	Partial combustion on top edge	Sustained combustion, entire sample burned				
Burn Rate	(inch/min)		5.9	8.2	Not measured		7.5				7.2	6.4			6.3				5.3
Burn	cm/min		15	20.8	Not m		19				18.3	16.3			16.0				7.8
locity	(ft/sec)	170.5	170.5	85.2	196	144.5	96.3	299.5	249.5	220.6	194.4	169	204	188	256	341	308	396	147
Air Velocity	m/sec	52	52	26	59.7	44	29.4	91.3	76	67	29	51.5	62.2	57.3	78	104	93.9	120.7	44.8
Air Flow	(lbm/sec)	1.0	1.0	0.5	2.0	1.5	1.0	3.0	2.5	2,25	2.0	2.5	3.0	2.75	1.5	2.0	1.75	2,25	1.5
Air	Kg/sec	0.45	0.45		0.91	0.68	0.45	1.36	1.13	1.02	0.91	1.13	1.36	1,25	0.68	0.91	0.79	1.02	0.68
Pressure	(psia)	65	65	64	113	115	115	111	111	113	114	164	163	162	65	65	63	63	113
Pre	КРа	448	448	441	779	793	793	765	765	779	786	1131	1124	1117	448	448	434	434	779
Test #		SAPOI	8AP02	8AP03	8AP04	8AP05	8AP06	8AP07	8AP08	8AP09	8AP10	8AP11	8AP12	8AP13	8AQ14	8AQ15	84016	8AQ17	8AQ18

Sample:  $2.54~\mathrm{cm}~\mathrm{x}~7.62~\mathrm{cm}~\mathrm{x}~0.06~\mathrm{cm}~\mathrm{Titanium}$  Temperature:  $399^{\circ}\mathrm{C}$ 

TABLE 4

TITANIUM COMBUSTION TEST RESULTS, SAMPLE B

15-13					
Summary of Test	Slight combustion on top edge Slight combustion on top corners, however, igniter was 1/4" low Igniter fused to sample, not a good ignition Sustained combustion on leading edge; however, entire sample did Sustained combustion, entire sample burned	Partial combustion on top edge, however, igniter was 1/4" low Sustained combustion; however, entire sample did not burn Partial combustion on leading corner and top edge Sustained combustion; however, entire sample did not burn Igniter went over the top and did not ignite the sample	Sustained combustion, entire sample burned Sustained combustion, entire sample burned Ignition was marginal, slight combustion on top edge Slight combustion on top edge Partial combustion on top edge	Sustained combustion, entire sample burned Sustained combustion, entire sample burned Slight combustion on trailing edge; however, igniter was too low Igniter only burned partially Slight combustion on top edge	Slight combustion on top edge; however, most of igniter did not hit the sample Sustained combustion, entire sample burned Partial combustion on top edge
Burn Rate in (inch/min)	5.			3.3	3.1
Burn cm/min	9.9		14.7	10.7	7.9
Velocity (ft/sec)	97 58 60 50 49	128 112 111 90 56	79 79 118 118	79 79 98 100 80	0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0
Air Ve m/sec	30 18 18 15	334 34 27 17	24 36 36 30	24 30 40 24	18 15 18
Air Flow Ka/sec (lbm/sec)	1.0	2.3 2.0 2.0 1.6	1.4	2.0	0.0
Air Ka/sec	0.45 0.27 0.23 0.23	1.04 0.91 0.73 0.45	0.64 0.64 1.36 1.13	0.91 0.91 1.13 0.45	0.27
Pressure		117 116 117 115	115 115 165 165	165 165 165 65	65 65 65
Pre	461 461 448 448	806 800 806 793 793	793 793 1138 1138 1131	1138 1138 1138 448	448 448 448
Test #	8AH02 8AH03 8AI01 8AI02	8AI04 8AI05 8AI06 8AI07 8AI08	8AJ01 8 8AJ02 8AJ03 8AJ04 8AJ05	8AJO6 8AJO7 8AJO8 8ASO9 8ASIO	8AS11 8AS12 8AS13

Temperature: 121°C ±2°C

Sample: 2.54 cm x 7.62 cm x 0.16 cm Titanium

TABLE 5

TITANIUM COMBUSTION TEST RESULTS, SAMPLE B

Summary of Test		Slight combustion on top edge	top	top	ent	Ignition was marginal, sample did not burn	Sustained combustion, entire sample burned	Ignition was marginal, sample did not burn	Slight combustion on top edge	Sustained combustion, entire sample burned	Slight combustion on top edge	Sustained combustion, entire sample hurned	Ignition was marginal, sample did not ignite	Sample installed in reverse direction	Sustained combustion, entire sample burned	Slight combustion on top edge	Slight combustion on top edge	top	ed combustion,	Slight combustion on top edge	Sustained combustion, entire sample burned	Bad ignition	Bad ignition	Slight combustion on top leading edge	Sustained combustion, entire sample burned	Slight combustion on top edge	
Burn Rate	(inch/min)				3.18		4.38			4.74		5.34			2,88				5.18		Not measured				4.73		
Burr	cm/min				8.0		11.1			12.0		13.6			7.3				13.2		Not me				12.0		
locity	(ft/sec)	405	275	203	108	344	153	344	233	78	189	107	135	54	54	150	86	98	49	86	188	188	159	159	126	157	
Air Velocity	m/sec	123	84	62	33	105	47	105	71	24	57	32	41	16	16	46	30	30	15	30	57	57	48	48	38	48	
Flow	(lbm/sec)	3.0	2.0	1.5	0.8	4.5	2.0	4.5	3.0	1.0	3.5	2.0	1.0	0.4	0.4	1.5	1.0	1.0	0.5	1.0	2.5	3.0	2.5	2.5	2.0	2.5	
Air	Kg/sec	1.36	0.91	0.68	0.36	2.04	0.91	2.04	1.36	0.45	1.59	0.91	0.45	0.18	0.18	0.68	0.45	0.45	0.23	0.45	1.13	1.36	1.13	1.13	0.92	1.13	
Pressure	(psia)	65	64	65	65	115	115	115	113	113	163	165	65	65	65	88	06	90	06	90	117	140	141	142	140	140	
Pre	кРа	448	441	448	448	793	793	793	779	779	1124	1138	448	448	448	607	621	621	621	621	807	965	972	979	965	965	
Test #		8AE01	8AE02	8AE03	8AE04	8AE05	8AE06	8AE07	8AE08	8AE09	8AE10	8AE11	8AF01	8AF02	8AF03	8AF04	8AF05	8AF06	8AF07	8AF08	8AF09	8AF10	8AF11	8AF12	8AF13	8AG01	

Sample: 2.54 cm x 7.62 cm x 0.16 cm Titanium Temperature:  $260^{\circ}\text{C}$ 

TABLE 5 (Cont'd)

Test #	Pre	Pressure	Air	Air Flow	Air V	Air Velocity	Burr	Burn Rate	Summary of Test	
	kРа	(psia)	Kg/sec	Kg/sec (lbm/sec)	m/sec	m/sec (ft/sec)	cm/min	cm/min (inch/min)		
8AG02	1138	165	1.13	2.5	41	133	13.3	5.22	Sustained combustion	
8AG03	1138	165	1.36	3.0	49	160			Slight combustion on top edge	
8AG04	1151	167	1,36	3.0	48	158			Slight combustion on top edge	
8AN01	793	115	0.91	2.0	47	153			Slight combustion on trailing edge	
8AN02	786	114	0.45	1.0	24	77	10.2	4.01	Sustained combustion, entire sample burned	burned
8AN03	793	115	0.68	1.5	35	115	11.9	4.7	Sustained combustion, entire sample burned	burned
8AN04	779	113	0.79	1.75	41	136			Partial combustion	
8AN05	772	112	0.91	2.0	48	157			Slight combustion on top edge	
8AS06	448	65	0.36	0.8	33	108	7.6	3.0	Sustained combustion, 80% of sample burned	burned
8AS07	448	65	0.45	1.0	41	135	5.1	2.0	Sustained combustion, 80% of sample burned	burned
8AS08	448	65	0.50	1.1	45	149			Slight combustion on trailing edge	

TABLE 6

TITANIUM COMBUSTION TEST RESULTS, SAMPLE B

Summary of Test		Partial combustion on trailing edge top corner	Slight combustion on top edge	Partial combustion on top edge	Sustained combustion, entire sample burned	Partial combustion on top edge	Sustained combustion, entire sample burned	Partial combustion on top edge and trailing edge	Sustained combustion, entire sample burned	Partial combustion on top edge	Sustained combustion, entire sample burned	Partial combustion on top edge and trailing edge	Sustained combustion, entire sample burned	Sustained combustion, entire sample burned	Slight combustion on top edge	Sustained combustion, entire sample burned	Partial combustion on top edge and trailing edge
Burn Rate	cm/min (inch/min)				4.8		0.9		4.65		4.9		3.5	3.6		3.2	
Bur	cm/min				12.2		15.2		11.8		12.4		8.9	9.1		8.1	
Air Velocity	(ft/sec)	135	157	130	138	165	149	149	117	154	134	156	100	135	203	169	209
Air Ve	m/sec	41	48	40	42	20	45	45	36	47	41	92	30	41	62	52	64
Flow	(lbm/sec)	1.0	3.0	2.5	2.5	3.0	2.75	2.75	1.5	2.0	1.75	2.0	0.75	1.0	1.5	1.25	1.5
Air	Kg/sec	0.45	1.36	1.13	1.13	1.36	1.25	1.25	0.68	0.91	0.79	0.91	0.34	0.45	0.68	0.57	0.68
Pressure	(psia)	65	168	169	163	160	162	162	113	114	115	113	99	.69	65	65	63
Pre	кРа	448	1158	1165	1124	1103	1117	1117	779	786	793	779	455	448	448	448	434
Test #		8AH01	8AG05	8AG06	8A201	8AQ02	8AQ03	8AQ04	8AQ05	8AQ06	8AQ07	8A208	8AQ09	8AQ10	8AQ11	8AQ12	8AQ13

Sample: 2.54 cm x 7.62 cm x 0.16 cm Titanium Temperature:  $399^{\circ}$ C

TABLE 7

TITANIUM COMBUSTION TEST RESULTS, SAMPLE C

Summary of Test	Slight combustion on top edge	Partial combustion on leading edge	Ignition failure	Partial combustion on leading edge	Sustained combustion, sample completely burned	Sustained combustion, sample completely burned	Sustained combustion, sample completely burned	Slight combustion on trailing and top edges	Sustained combustion, sample completely burned	Sustained combustion, sample completely burned
Burn Rate cm/min (inch/min)					11.7 4.6	11.2 4.4	12.7 5.0		15.2 6.0	Not measured
Air Velocity m/sec (ft/sec)	100	80	20	20	5.7	114	114	201	174	40
Air Ve m/sec	30	24	15	15	17	35	44	61	53	12
Flow (lbm/sec)	1.0	0.8	0.5	0.5	1.0	2.0	2.5	3.5	3.0	0.4
Air Kg/sec	0.45	0.36	0.23	0.23	0.45	0.91	1.10	1.60	1.40	0.22
(psia)	65	65	65	64	115	114	113	113	112	65
Pressure kPa (psia	448	448	448	441	793	786	779	779	772	448
Test #	SALO1	8AL02	8AL03	8AL04	8AL05	8AL06	8AL07	8AL08	8AL09	8AL10

Sample: TF-39 6th Stage Compressor Blade Temperature:  $121^{\circ}\text{C}$  (250°F)

TABLE 8

TITANIUM COMBUSTION TEST RESULTS, SAMPLE C

Summary of Test	Sample incorrectly installed	Not a good ignition	Sustained combustion, sample completely burned	Sustained combustion, sample completely burned	Sustained combustion	Slight combustion on trailing and top edges	Sustained combustion, sample completely burned	Sustained combustion, sample burned about 50%	Slight combustion on trailing edge	Sustained combustion, sample completely burned
Burn Rate cm/min (inch/min)			4.9	5.4	7.9		4.7	3.0 .		5.3
Burn cm/min			12,4	13.7	20.0		11.9	7.6		13.5
Air Velocity sec (ft/sec)	137	173	65	115	154	195	195	235	235	147
Air V m/sec	42	42	20	35	47	59	69	72	72	45
Flow (lbm/sec)	1.0	1.0	0.5	1.5	2.0	2.5	2.5	3.0	3.0	1.5
Air Kg/sec	0.45	0.45	0.23	0.68	0.91	1.1	1.1	1.4	1.4	0.68
Pressure Pa (psia)	64	64	68	115	114	113	113	112	112	113
Pres	441	441	469	793	786	779	779	772	772	779
Test #	8A001	8A002	8A003	8A004	8A005	84006	8A007	84008	8A009	8AQ18*

Sample: TF-39 6th Stage Compressor Blade

Temperature:  $260^{\circ}\text{C}$  ( $500^{\circ}\text{F}$ ) except \* which was  $399^{\circ}\text{C}$  ( $750^{\circ}\text{F}$ )

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TABLE 9

TITANIUM COMBUSTION TEST RESULTS, SAMPLE D

	Test #	Pres	Pressure Pa (psia)	Air Kg/sec	Air Flow sec (lbm/sec)	Air m/sec	Air Velocity sec (ft/sec)	Burn cm/min	Burn Rate in (inch/min)	Summary of Test
	Tempera	ture =	Temperature = 149 <sup>o</sup> C (300 <sup>o</sup> F)	300°F)						
	8AR05	517	75	0.45	1.0	28	8 6	0	-	Ignition on leading edge, only small amount of burning
	8AR06 8AR07	531	77	0.45	1.0	27	06	T . O T	₽	Ignition was not good
	8AR08	531	77	0.34	0.75	10	89	12.2	4.6	Ignition on leading edge, sample burned completely
	Tempera	Temperature =	260°C (500°F)	500 <sup>°</sup> F)						
	8ARO9 8ARIO	517	75	0.64	1.4	50	164	11.4	4.5	Ignition on trailing edge, sample burned completely Ignition was not good, molten Ii did not hit sample remition on leading edge, only small amount of burning
	8AR11 8AS01	517	75	0.45	1.0	36	164	10.4	4.1	Ignition on trailing edge, sample burned completely
21.	8AS02	524	76	0.34	0.75	26	87	15.2	0.9	Ignition on leading edge, sample burned completely
	Tempera	ature =	Temperature = $371^{\circ}$ C (700 $^{\circ}$ F)	700 <sup>O</sup> F)						
	8AS03	855	124	0.77	1.7	44	146	13.5	5,3	trailing edge, sample burned
	8AS04 8AS05	862	125	0.77	1.7	4 4 4 4	144	10.2 Not me	0.2 4.0 Not measured	Ignition on leading edge, sample burned completely Ignition on leading edge, sample burned completely
					Sample:	TF-30 6th	6th Stage C	Stage Compressor Blade	Blade	Temperature: As indicated above Test No.

TABLE 10

MAXIMUM AIR VELOCITY FOR SUSTAINED COMBUSTION

PRES	PRESSURE	SAMPLE		MAXIMUM	MAXIMUM VELOCITY FOR SUSTAINED COMBUSTION	SUSTAINED	COMBUSTION	
kPa	(psia)		121°C	(250°F)	260°C	(500°F)	399°C	(750°F)
			m/sec	ft/sec	m/sec	ft/sec	m/sec	ft/sec
448	(65)	A (.06 cm)	52	(170)	47	(153)	80	(262)
793	(115)	A	51	(167)	20	(165)	63	(207)
1138	(165)	A	43	(141)	49	(161)	54	(178)
484	(65)	B (.16 cm)	17	(55)	43	(142)	52	(169)
798	(115)	В	34	(112)	40	(131)	44	(144)
1138	(165)	В	27	(88)	44	(145)	45	(149)
448	(65)	C (TF-39 6th)	14	(45)	>20	(>9<)	Not Tested	sted
793	(115)	Ü	57	(188)	99	(215)	45	(>147)
1138	(165)	E (β alloy)	Not Tes	Tested	20	(99)	Not Tes	Tested

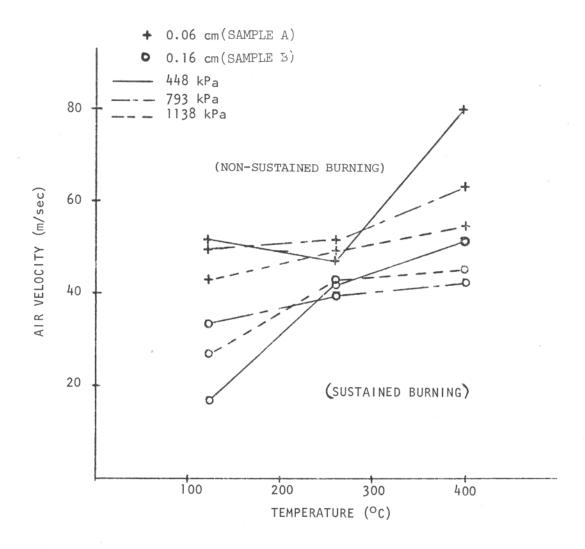


Figure 13. Sustained, Non-Sustained Burning Data for Samples A and B

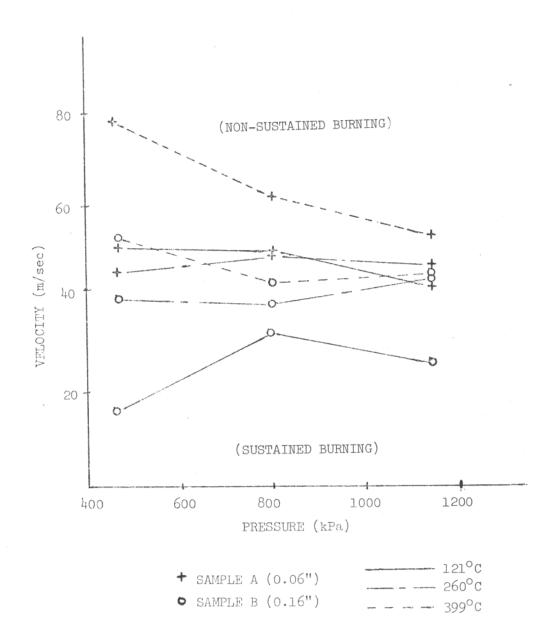


Figure 14. Sustained, Non-Sustained Burning Data for Samples A and B

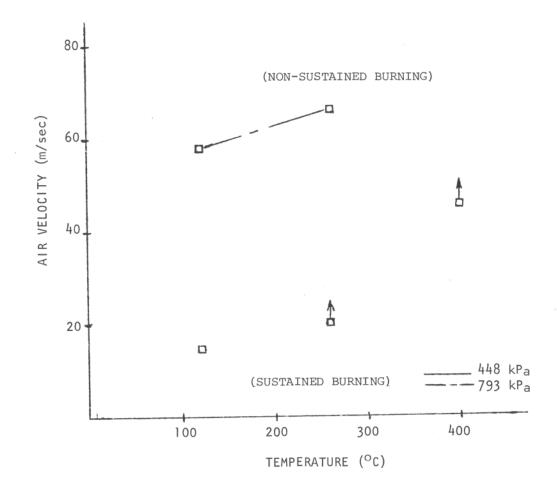


Figure 15. Sustained, Non-Sustained Burning Data for Sample C

also pressure. The pressure effect may be caused by a change in the air flow across the sample at higher pressures.

There are not sufficient data points with Sample D to establish any trends in the burning characteristics. The data on both types of blades indicate that the blades will burn generally in the same air flow conditions as the flat surface samples.

A few tests were conducted with a  $\beta$  structure titanium alloy (Sample E). The thickness of this sample was between that of Sample A and Sample B. It was not possible to achieve sustained burning with this sample at an air velocity that supported sustained combustion with Samples A and B. In addition, the burn rate of this sample at 260°C, 1138 kPa, and 20 m/sec air velocity is 10 cm/min, which is somewhat less than the burn rate measured for Samples A and B at the same air flow conditions. The burn rate of the other samples is discussed in the next section.

These limited test results indicate that the alloy and structure of titanium have a definite effect on the burning characteristics.

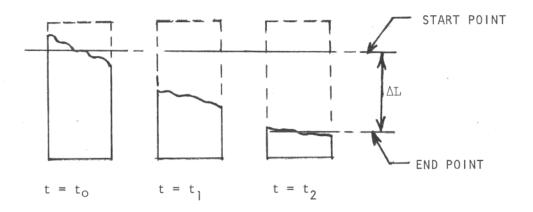
## SECTION IV

## BURN RATE ANALYSIS

## A. ANALYSIS DESCRIPTION

The burning rate for samples which exhibited sustained burning is determined by analyzing the high speed motion pictures. Each test was photographed at 60 frames per second. Burning rate can be expressed in several ways, such as weight change or volume change. The speed of the burning edge as it burned down the sample is used for this analysis. Difficulties occur, however, with this approach because the sample often does not burn at a constant rate over the whole sample. What is described here is a determination of an average burn rate on the portion of the sample which did experience a fairly steady and even burn rate. The top edge near the igniter and the bottom edge near the sample holder were excluded. As illustrated in Figure 16, the burn rate is determined by measuring the time it takes the sample to burn between two points on the sample. The end points were varied somewhat from test to test to exclude variations such as an unusually slow initial burn rate or a burn pattern which developed an odd shape. If less than 2 cm length of sample burning could not be approximated by a straight line burn pattern, the burn rate was not determined for that specific test.

Analysis of the motion picture film reveals some general characteristics of the burning. If the sample ignites across the entire top edge, it generally burns from top to bottom. Sometimes the sample will burn faster down the front edge and then burn back toward the trailing edge. Other



BURN RATE = 
$$\frac{\Delta L}{t_2 - t_0}$$

Figure 16. Burn Rate Measurement Technique

tests result in the sample burning faster down the trailing edge and then forward into the sample.

## B. RESULTS AND DISCUSSION

The burn rate data are tabulated in Tables 11 through 14. Although the burn rate varies considerably even at the same air flow conditions, it is possible to establish trends and the lack of trends from these data.

The temperature effect on the burn rate is too small to be determined. Not much effect is expected because the ambient temperature change is small compared to the temperature at the burning surface.

A definite trend due to pressure is apparent with Sample B and somewhat apparent with Sample A. The samples burn faster at higher pressure. The air is more dense at the higher pressure and thus more oxygen is available for combustion.

In general, Sample A burned faster than Sample B under the same test conditions.

The burn rate of the compressor blades (Samples C and D) does not show a trend due to pressure or temperature. Since the blades were not tested over the full temperature and pressure range, it is difficult to determine a trend. The blades appear to generally burn at a rate between the rates of Samples A and B. The blade thickness varies from the thin edges, which are approximated by Sample A, to the thick center, which is approximated by Sample B. The measured burn rate for the blades is an average rate over the complete blade, and thus would be expected to be within this range. In a few tests, a blade edge burned several times faster than the center portion.

TABLE 11

AVERAGE BURN RATE DATA SUMMARY, SAMPLE A

URN RATE inch/min	6.3 7.0 3.7	7.2	5.2	6.6 8.3 6.7	6.69	6.887	7.5	ή*9
AVERAGE BURN RATE cm/min inch	16 18 9	18	13	1.7 2.1 1.7	20 24 17	15 21 16	19 18 14	16
JOCITY ft/sec	107 162 162	154	203	115 134 156	107 134 161	170 85 .256	96 194 147	169
AIR VELOCITY m/sec ft	33 64 64	7.4 7.4	62	35 41 48	33 41 49	52 26 78	29 59 45	52
PRESSURE psia	6 6 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	115	65	115 115 115	165 165 165	655	115	165
PRES KPa	7778 7778 7778 7778	793	844	793 793 793	1138 1138 1138	7778 778 778	793 793 793	1138
TEMPERATURE OC OF	250 250 250	250	500	500	500	750 750 750	750 750 750	750
TEMPER	121 121 121	121	260	260 260 260	260 260 260	399 399 399	399 399 399	399

SAMPLE A (.06 CM thick)

TABLE 12

AVERAGE BURN RATE DATA SUMMARY, SAMPLE B

AVERAGE BURN RATE cm/min inch/min	2.6	3.9	4.2 3.3	8.6.00 8.00 8.00 8.00	1.4 1.4 7.4 1.0	5.5		4.7 4.9	4.8 6.0
AVERAGE   cm/min	L-80	15	111 8	2000	11 11 11 11 11 11 11 11 11 11 11 11 11	13	0,0,0	12	12
COCITY ft/sec	760 2	79	79	108 54 108 135	153 78 77 115	107	100	137	138
AIR VELOCITY m/sec ft/	15	24	27	33 16 33 41	47 24 25 35	33	31 41 52	36	45 45
SURE psia	65	115	165 165	6555	115 115 115	165	655	115	165
<u>PRESSURE</u> KPa psia	448 65 448 65	793 115 793 115	1138 165 1138 165		793 115 793 115 793 115 793 115	1138 165 1138 165	448 65 448 65 65 65	793 115 793 115	1138 165 1138 165 thick)
PRESSURE									1138 1138 (0.16 CM thick)
PRESSURE			1138	877 877 877		1138	877 877 877		1138 1138 0.16 CM thick)

TABLE 13

AVERAGE BURN RATE SUMMARY, SAMPLE C

-15-13					
AVERAGE BURN RATE min inch/min	6.5	4.6 5.0 6.0	5.0	5.4 7.9 4.7 3.0	
AVERAGE cm/min	16.5	12 11 13	13	14 20 12 7.6	13.5
AIR VELOCITY sec ft/sec	04	57 1114 1114 174	65	115 154 195 235	147
AIR V m/sec	12	17 35 44 53	50	35 47 72	5 †7
URE psia	65	115 114 113 112	68	115 114 113 112	113
PRESSURE kPa	844	793 786 779	69†	793 786 779	611
TEMPERATURE OC	250	250 250 250 250	200	500 500 500	750
TEMPE	121	121 121 121 121	560		399

SAMPLE C: (TF-39 6th Stage Blade)

TABLE 14

AVERAGE BURN RATE SUMMARY; SAMPLE D

17-1-					
RN RATE	inch/min	4.1	4.5 4.1 6.0	3.6	5.3
AVERAGE BURN RATE	cm/min	10.4	11.4 10.2 15	9.1	13.5
AIR VELOCITY	ft/sec	089	164 164 87	153	144 144
AIR VE	m/sec	27	200	747	†; †; †; †;
PRESSURE	psia	77	75 75 76	11.4	124
PRE	кРа	531 531	517 517 524	786	855 862
TEMPERATURE	보이	300	500	200	700
TEMPER	Do	149 149	092	560	371 371

SAMPLE D: (TF-30 6th Stage Blade)

## SECTION V

## FLAME SUPPRESSION TESTS

## A. TEST DESCRIPTION

The titanium flame extinguishing tests were conducted to determine what concentration of argon gas is required to extinguish a titanium fire. Argon is believed to be inactive in the presence of the high temperature titanium flame and can thus be used to decrease the oxygen concentration to a level below that required to support combustion. The tests were conducted at air flow conditions of 260°C (500°F) and 1138 kPa (165 psia). The air velocity past the blade prior to argon injection was 24 m/sec (80 ft/sec). The temperature and density of the mixture changed the velocity in the test chamber after injection of the argon gas, but these changes were not sufficient to cause the flame to extinguish. The mixture velocity after argon injection dropped to 19 m/sec (61 ft/sec) for the worst case (100% argon).

Attempts to calibrate the concentration of argon by using a rotary vane flow meter did not prove successful because the meter output did not change sufficiently over the range of use in these tests and thus caused inaccurate data. With a sufficiently higher pressure in the argon manifold than in the test chamber, the flow is held constant by a critical orifice located after the solenoid operated injection valve. After a short transient upon opening the valve, the flow is controlled by the argon manifold pressure which also experiences a short transient (less than 0.1 sec) and then reaches a steady state.

The following procedure was used to calibrate the argon injection system. The test chamber was operated at the air flow conditions required for the flame extinguishment tests. Argon was injected at six different values of manifold pressure, and the argon-air mixture was sampled near the test sample at these six values. The values of pressure required were estimated from calculations of critical orifice flow. The range of argon concentration of interest was between 50% and 100%. The six gas mixture samples were analyzed for composition. The percent argon mixture can thus be determined from the argon manifold pressure. The calibration data are illustrated in Figure 17. The amount of argon injected is assumed to be linear with respect to the manifold pressure. The highest pressure data point does not fall in line with the other points. Most likely, an excess of the 100% argon is actually injected at this pressure and the excess is bled off through the pressure regulating bleed valve.

The argon-air mixture reaches a quasi steady-state temperature within a few seconds following injection of the argon. The argon cools after expansion through the critical orifice. The pipe between the injection point and the test chamber is, however, at the initial air flow temperature and contains a large mass which heats the argon-air mixture. The temperature of the mixture did not change significantly after reaching the steady-state value.

## B. RESULTS AND DISCUSSION

The test results are tabulated in Table 15. Four tests were conducted with Sample B, and the flame was extinguished in each test. Seven tests were conducted with the thinner sample (Sample A) with the argon concentration ranging from 45% to 100%.

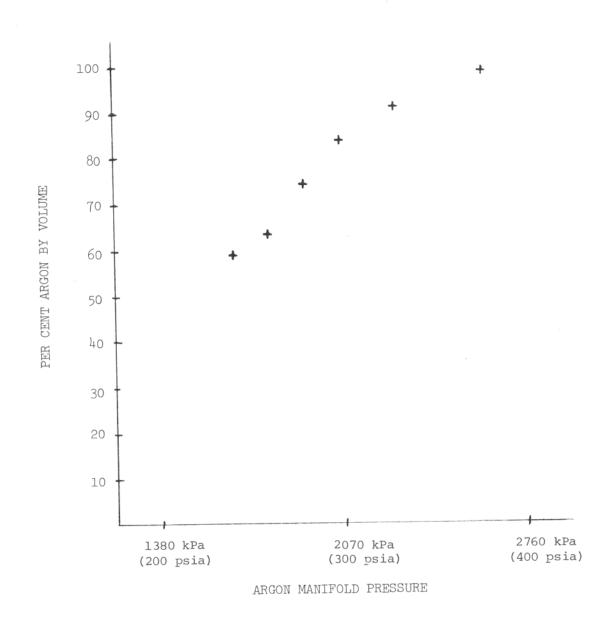


Figure 17. Argon Injection Calibration Curve

TABLE 15

ARGON GAS EXTINGUISHING DATA

TEST	SAMPLE	% ARGON	TIME TO EXTINGUISHMENT
8AZ01	В	90	40.1 sec
8AZ02	В	98	<0.1 sec
8AZ04	В	100	<0.1 sec
8AZO4	В	100	< 0.1 sec
8avo8	В	48	1.0 sec
8AV10	В	62	0.5 sec
8AV11	В	50	0.5 sec
8AZ05	A	100	< 0.1 sec
8AZ06	A	90	<.0.1 sec
8AZ07	A	73	0.15 sec
8AZ08	A	66	0.15 sec
8AZ09	A	55	7 sec (See text)
8AZ10	A	45	Continued to burn
8AZ12	A	100	< 0.1 sec

Pressure 1138 kPa (165 psia)
Temperature 260°C (500°F)

Air Velocity 24m/sec (80 ft/sec) prior to argon injection

It is somewhat difficult to determine exactly when the flame is extinguished since the material remains hot for a period of time. The argon arrival time at the sample was calculated by assuming that the argon and air were completely mixed at the injection point and that steady state conditions were reached immediately after the valve opened. The flame was considered extinguished when all molten material stopped leaving the sample surface and activity on the surface slowed considerably. On the tests with high argon concentration, this effect took place within 0.1 second after the argon mixture reached the sample; however, the time to extinguishment listed in Table 15 can only be considered accurate to within 0.1 second due to the interpretive nature of the data analysis.

Concentrations of argon above 65% effectively extinguished the flame, although slightly longer times are required for the tests at 66% and 73% than for the tests with 90% and above. The test conducted at 55% concentration showed a different characteristic. A definite slowing of the combustion took place within 0.1 second after arrival of the argon; however, the sample continued to burn for another 8 seconds before surface activity stopped and the sample cooled. The test conducted with Sample A at 45% argon concentration showed the same initial decrease in burning activity; but in this test, the burning continued for considerable time and the sample almost completely burned.

Additional tests were conducted to determine what would happen if the argon concentration were decreased after initially extinguishing a burning sample but prior to the sample cooling. In these tests, the chamber air flow conditions were the same as shown in Table 15. The test sample (Sample A) was ignited and allowed to achieve a steady burn. The flame was then

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